

Preliminary communication

Chelated bis-silyl-platinum complexes

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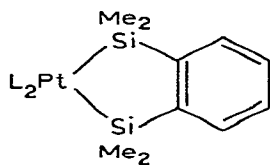
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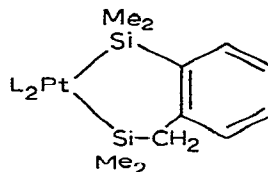
SUMMARY

The preparations are described of some bis-silyl-platinum complexes in which the Si-Pt-Si linkage forms part of a 4-, 5- or 6-membered ring.

The dihydrides o -(HMe₂Si)₂C₆H₄ and o -(HMe₂Si)C₆H₄CH₂SiMe₂H have been found to react with [PtL₂(C₂H₄)] (L = PPh₃) at room temperature to give the 5- and 6-membered cyclic bis-silyl complexes (I), m.p. 146-149° (dec.) and (II), m.p. 142-146°,



(I)



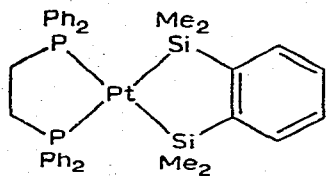
(II)

respectively. The disiloxane (HPh₂Si)₂O analogously gives a 4-membered cyclic species [Pt(SiPh₂OSiPh₂)L₂], m.p. 166-170°, at 45°; (HMe₂Si)₂O, however, gives only the hydrido(silyl) complex, *cis*-[PtH(SiMe₂OSiMe₂H)L₂], m.p. 100-104°, at 45°, but the latter cyclizes at 75° to give [Pt(SiMe₂OSiMe₂)L₂], m.p. 172-174°. The dihydrides o -(HMe₂SiCH₂)C₆H₄ and HMe₂Si(CH₂)₄SiMe₂H, which could in principle give complexes involving 7-membered rings, in fact give only the hydrido(silyl) complexes, *cis*-[PtH{SiMe₂CH₂C₆H₄(CH₂SiMe₂H)-*o*}L₂], m.p. 75-78°, and *cis*-[Pt{SiMe₂(CH₂)₄SiMe₂H}L₂], m.p. 110-112°, which do not cyclize at 75°.

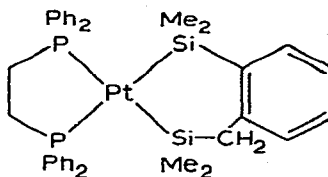
The compounds analogous to (I) and (II), but with PMePh₂ (L') in place of PPh₃ ligands, are readily formed on treatment of either [Pt(PMePh₂)₄] (neat) or [Pt(PMePh₂)₂(CO₃)] (in benzene) with the appropriate dihydrides at room temperature.

The disiloxane $(\text{HMe}_2\text{Si})_2\text{O}$ gives the cyclic $[\text{Pt}(\text{SiMe}_2\text{OSiMe}_2)\text{L}'_2]$ directly on reaction with $[\text{Pt}(\text{PMePh}_2)_4]$ in benzene under reflux or with $[\text{Pt}(\text{PMePh}_2)_2(\text{CO}_3)]$ in the same solvent at room temperature.

Compounds (I) and (II) react readily with 1,2-bis(diphenylphosphino)ethane to give the doubly-chelated species (III), m.p. $263-266^\circ$, and (IV), m.p. $246-248^\circ$, respectively. Compounds (I) and (II) also react readily with 2 molar proportions of bromine, methyl iodide, or phenylacetylene with cleavage of both Si—Pt bonds; in the case of (I) [(II) was not examined in this respect], use of only 1 molar proportion of the reagents leads to the same products along with unchanged (I), indicating that once the first Si—Pt has been broken the second is cleaved markedly more rapidly.



(III)



(IV)

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